# SAFETY DATA SHEET

national diagnostics

Conforms to regulation (EC) no. EU 453/2010

## SECTION 1 - IDENTIFICATION OF THE SUBSTANCE/MIXTURE AND OF THE COMPANY/UNDERTAKING

#### 1.1 Product Identifier

Product Name: Denaturation Solution Product Number: EC-875

## 1.2 Relevant Identified Uses of the Substance/Mixture and Uses Advised Against

Investigational research by professional users

## 1.3 Details of the Supplier of the Safety Data Sheet

#### Manufacturer

National Diagnostics 305 Patton Drive Atlanta, GA 30036 (404) 699-2121 (800) 526-3867 info@nationaldiagnostics.com

# 1.4 Emergency Telephone Number

#### Chemtrec

1-800 424-9300 (U.S. & Canada) 01-703-527-3887 (outside U.S. & Canada)

#### **SECTION 2 - HAZARDS IDENTIFICATION**

#### 2.1 Classification of the Substance or Mixture

## Classification according to Regulation (EC) No. 1272/2008 [EU-GHS/CLP]

H290 - Corrosive to Metals

H314 - Skin Corrosion/Irritation (Category 1B)

## 2.2 Label Elements

# **GHS LABEL ELEMENTS AND CLASSIFICATION**

#### **GHS Label Elements**



#### DANGER

H290 - May be corrosive to metals.

H314 - Causes severe skin burns and eye damage.

 ${\tt P260-Do\ not\ breathe\ dust/fumes/gas/mist/vapors/spray}.$ 

 ${\tt P280-Wear\ protective\ gloves/protective\ clothing/eye\ protection/face\ protection}.$ 

P302+P361+P353 - IF ON SKIN (or hair): Remove/Take off immediately all

contaminated clothing. Rinse SKIN with water/shower.

P305+P351+P338 - IF IN EYES: Rinse cautiously with water for several minutes.

Remove contact lenses if present and easy to do. Continue rinsing.

## 2.3 Other Hazards

None found.

# **SECTION 3 - COMPOSITION/INFORMATION ON INGREDIENTS**

## 3.2 Mixture

#### **Chemical Names/Description**

Aqueous solution of sodium chloride and sodium hydroxide.

## **Component List**

| Component        | % Comp. | CAS#      | EC#       | 1278/2008<br>Classification |
|------------------|---------|-----------|-----------|-----------------------------|
| Sodium Hydroxide | 2.0     | 1310-73-2 | 215-185-5 | H290, H314                  |
| Sodium Chloride  | 8.8     | 7647-14-5 | 231-598-3 | N.A.                        |

# **SECTION 4 - FIRST AID MEASURES**

# 4.1 Description of First Aid Measures

#### Inhalation

Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Call a physician.

## Ingestion

Do not induce vomiting. If swallowed and the person is conscious, immediately give large amounts of water. Get medical attention.

#### Skin

Immediately flush skin with plenty of soap and water for at least 15 minutes while removing contaminated clothing and shoes. Get medical attention. Wash clothing before reuse. Thoroughly clean shoes before reuse.

#### Eyes

Immediately flush eyes with plenty of water for at least fifteen minutes, lifting lower and upper eyelids occasionally. Get medical attention immediately.

# 4.2 Most Important Symptoms and Effects, Both Acute and Delayed

#### Inhalation

#### Sodium Hydroxide:

Sneezing, sore throat or runny nose.

#### Sodium Chloride:

No information found.

#### Ingestion

#### Sodium Hydroxide:

Bleeding, vomiting, diarrhea, fall in blood pressure. Symptoms may appear days after exposure.

#### Sodium Chloride:

No information found.

#### Skin

#### Sodium Hydroxide:

Redness, pain, burns.

#### Sodium Chloride:

No information found.

#### **Eyes**

## Sodium Hydroxide:

Redness, pain, tearing.

#### Sodium Chloride:

No information found.

## 4.3 Indication of Any Immediate Medical Attention and Special Treatment Needed

Unknown/not applicable

# **SECTION 5 - FIRE FIGHTING MEASURES**

## 5.1 Extinguishing media

Use media appropriate to the primary cause of fire.

#### 5.2 Special Hazards Arising from the Substance/Mixture

# **Hazardous Combustion Products**

N.A.

#### **Hazardous Decomposition Products**

No hazardous decomposition products.

## **Hazardous Polymeriation**

Will not occur under normal conditions of use (See Sections 10.4 & 10.5).

#### 5.3 Advice for Firefighters

In the event of a fire, wear full protective clothing and NIOSH-approved self-contained breathing apparatus with full facepiece operated in the pressure demand or other positive pressure mode.

#### 5.4 Further Information

No data available.

# **SECTION 6 - ACCIDENTAL RELEASE MEASURES**

# **6.1 Personal Precautions**

Wear appropriate protective equipment as specified in Section 8.

#### 6.2 Environmental Precautions

Prevent discharge into the environment. Dike spills and stop leakage where practical. Do not allow material to enter drains.

#### 6.3 Methods and Materials for Containment and Cleaning Up

Ventilate and isolate area. Prevent from entering floor drains. Residues can be diluted with water, neutralized with dilute acid. Absorb neutralized residues with clay, vermiculite, or other inert substance. Place in suitable disposal container.

#### 6.4 References to Other Sections

For disposal information, see Section 13. For Protective clothing and equipment, see Section 8.

## **SECTION 7 - HANDLING AND STORAGE**

#### 7.1 Precautions for Safe Handling

Avoid contact and inhalation. Do not get in eyes, on skin, on clothing. Wash thoroughly after handling. Always add the caustic to water while stirring; never the reverse.

## 7.2 Conditions for Safe Storage (including any incompatibles)

Keep in a tightly closed container, stored in a cooled, dry, ventilated area away from sources of heat, moisture, and incompatibilities. Protect from freezing.

# Incompatibles

#### Sodium Hydroxide:

Acids, aluminum, tin and zinc metals.

#### Sodium Chloride:

No incompatibility data found.

#### 7.3 Specific End Uses

Investigational research by professional users

## **SECTION 8 - EXPOSURE CONTROLS/PERSONAL PRECAUTIONS**

#### 8.1 Control Parameters

#### **Component: Sodium Hydroxide**

ACGIH Threshold Limit Value (TLV): 2 mg/m3 (TLV) OSHA Permissable Exposure Limit (PEL): 2 mg/m3

## **Component: Sodium Chloride**

ACGIH Threshold Limit Value (TLV): none established OSHA Permissable Exposure Limit (PEL): None established

#### 8.2 Exposure Controls

#### **Engineering Controls**

A system of local and/or general exhaust is recommended to keep employee exposures below the Airborne Exposure Limits. Local exhaust ventilation is generally preferred because it can control the emissions of the contaminant at its source.

#### **Respiratory Protection**

For conditions of use where exposure to the dust or mist is apparent, a full-face dust/mist respirator may be worn. For emergencies or instances where the exposure levels are not known, use a full-face positive-pressure, air-supplied respirator.

#### **Eye Protection**

Use chemical safety goggles and/or a full face shield where splashing is possible. Maintain eye wash fountain and quick-drench facilities in work area.

#### **Skin Protection**

Wear impervious protective clothing, including boots, gloves, lab coat, apron or coveralls, as appropriate, to prevent skin contact.

#### **SECTION 9 - PHYSICAL AND CHEMICAL PROPERTIES**

## 9.1 Information on Basic Physical & Chemical Properties

| a. Appearance                  | Clear, colorless solution | b. Odor  | None    |
|--------------------------------|---------------------------|--|---------|
| c. Odor Threshold              | N.A.                      | d. pH  | 14      |
| e. Melting/Freezing Point (°C) | -7                        | f. Boiling point (°C)                              | 106     |
| g. Flash Point (°C)            | N.A.                      | h. Evaporation Rate                                | Water   |
| i. Flammability                | N.A.                      | j. Upper/Lower Flammability or<br>Explosive Limits | N.A.    |
| k. Vapor Pressure              | Water                     | I. Vapor Density (Air = 1)                         | Water   |
| m. Relative Density            | 1.05                      | n. Water Solubility                                | Soluble |

o. Partition Coefficient Mixture p. Autoignition Temperature (°C)

n-octanol/water

q. Decomposition Temperature N.A. r. Viscosity No data available.

(°C)

t. Oxidizing Properties

not an oxidizer

#### **SECTION 10 - STABILITY AND REACTIVITY**

#### 10.1 Reactivity

Alkaline solution- will react violently with strong acids. May corrode metals

N.A.

#### 10.2 Chemical Stability

s. Explosive Properties

Stable under recommended conditions of use and storage.

#### 10.3 Possibility of Hazardous Reactions

Will not occur under normal conditions of use (See Sections 10.4 & 10.5).

#### 10.4 Conditions to Avoid

Heat, moisture, incompatibles.

#### 10.5 Incompatible Materials

#### Sodium Hydroxide:

Acids, aluminum, tin and zinc metals.

#### Sodium Chloride:

No incompatibility data found.

## 10.6 Hazardous Decomposition Products

No hazardous decomposition products.

#### **SECTION 11 - TOXICOLOGICAL INFORMATION**

## **Product LD50 Values**

Oral Rat LD50 (mg/kg)

No data.

#### Dermal Rabbit LD50 (mg/kg)

No data.

## Component Cancer List Status

|                  | NTP Carcinogen |             |               |
|------------------|----------------|-------------|---------------|
|                  | Known          | Anticipated | IARC Category |
| Sodium Hydroxide | No             | No          | None          |
| Sodium Chloride  | No             | No          | None          |

#### **Potential Health Effects**

# Inhalation

#### Sodium Hydroxide

Severe irritant. Effects from inhalation of mist vary from mild irritation to serious damage of the upper respiratory tract, depending on severity of exposure. Severe pneuomonitis may occur.

#### Sodium Chloride

May cause respiratory tract irritation.

#### Ingestion

#### Sodium Hydroxide

Corrosive! Swallowing may cause severe burns of mouth, throat, and stomach. Severe scarring of tissue and death may result.

## **Sodium Chloride**

Ingestion of large amounts may cause gastrointestinal irritation. Ingestion of large amounts may cause nausea and vomiting, rigidity or convulsions. Continued exposure can produce a coma, dehydration and internal organ congestion.

## Skin

# Sodium Hydroxide

Corrosive! Contact with skin may cause irritation or severe burns and scarring with greater exposures.

#### Sodium Chloride

May cause skin irritation.

# Eyes

#### Sodium Hydroxide

Corrosive! Causes irritation of eyes, and with greater exposures it can cause burns that may result in permanent impairment of vision, even blindness

#### Sodium Chloride

May cause eye irritation.

# Carcinogenicity

## Sodium Hydroxide

Not listed as a carcinogen by either NTP or IARC.

#### **Sodium Chloride**

Not listed by ACGIH, IARC, NIOSH, NTP or OSHA.

## Mutagenicity

## Sodium Hydroxide

No information found.

#### Sodium Chloride

See actual entry in RTECS for complete information.

# Reproductive Toxicity

#### . Sodium Hydroxide

No information found.

#### Sodium Chloride

No data available.

# **Teratogenic Effects**

#### Sodium Hydroxide

No information found.

## Sodium Chloride

No information found.

# **Routes of Entry**

# Sodium Hydroxide

Inhalation or ingestion.

## Sodium Chloride

No information found.

# **Target Organ Statement**

#### Sodium Hydroxide

Persons with pre-existing skin disorders or eye problems or impaired respiratory function may be more susceptible to the effects of the substance.

# Sodium Chloride

No information found.

# **SECTION 12 - ECOLOGICAL INFOMATION**

## 12.1 Toxicity

**COMPONENT: Sodium Hydroxide** 

|   | Vertebrates                         | Invertebrates                    | Algae                      | Microorganisms |
|---|-------------------------------------|----------------------------------|----------------------------|----------------|
| Aquatic Toxicity (ppm unless otherwise noted)                 | LC50 (48hrs, golden orfe) 189mg/L   | EC50 (48hr, daphnia)<br>40mg/L   | No data                    | No data        |
|   | Birds                               | Arthropods                       | Plants                     | Microorganisms |
| Terrestrial Environment Toxicity (ppm unless otherwise noted) | No data                             | No data                          | No data                    | No data        |
| COMPONENT: Sodium Chloride                                    |                                     |                                  |                            |                |
|   | Vertebrates                         | Invertebrates                    | Algae                      | Microorganisms |
| Aquatic Toxicity (ppm unless otherwise noted)                 | LC50 (96hrs, bluegill)<br>5480 mg/L | LC50 (48 hr, daphnia)<br>874mg/L | LC50 (120hrs) 2430<br>mg/L | No data        |
|   | Birds                               | Arthropods                       | Plants                     | Microorganisms |

No data

IC50 (7 days) 500-1950mg/kg soil No data

## 12.2 Persistence and Degradability

Sodium Hydroxide

No data

**Sodium Chloride** 

No data

## 12.3 Bioaccumulative Potential

Sodium Hydroxide

No data

**Sodium Chloride** 

No data

## 12.4 Mobility in Soil

Sodium Hydroxide

No data

**Sodium Chloride** 

No data

#### 12.5 Results of PBT and vPvB Assessment

Sodium Hydroxide

Mineral salt

**Sodium Chloride** 

Not PBT or vPvB

# 12.6 Other Adverse Effects

Sodium Hydroxide

None

Sodium Chloride

None

# **SECTION 13 - DISPOSAL CONSIDERATIONS**

## **13.1 Waste Treatment Methods**

Offer surplus or non-recyclable product to licensed disposal company. Disposal is subject to user compliance with applicable law and product characteristics at time of disposal. Dispose of packaging as product.

# **SECTION 14 - TRANSPORT INFORMATION**

|                            | ADR/RID                      | IATA                         | IMO                          | DOT                          |
|----------------------------|------------------------------|------------------------------|------------------------------|------------------------------|
| 14.1 UN Number             | 1824                         | 1824                         | 1824                         | 1824                         |
| 14.2 Shipping Name         | Sodium Hydroxide<br>Solution | Sodium Hydroxide<br>Solution | Sodium Hydroxide<br>Solution | Sodium Hydroxide<br>Solution |
| 14.3 Hazard Class          | 8                            | 8                            | 8                            | 8                            |
| 14.4 Packing Group         | III                          | III                          | III                          | III                          |
| 14.5 Environmental Hazards | N.A.                         | N.A.                         | N.A.                         | N.A.                         |
| 14.6 Special Precautions   | N.A.                         | N.A.                         | N.A.                         | N.A.                         |

## **SECTION 15 - REGULATORY INFORMATION**

# 15.1 Safety, Health and Environmental Regulations/Legislation Specific for the Substance/Mixture United States

**TSCA Regulatory Statement** 

All intentional ingredients are listed on the TSCA Inventory.

#### SARA 311/312 Hazard Categories

| Component        | Fire | Pressure | Reactivity | Acute | Chronic |
|------------------|------|----------|------------|-------|---------|
| Sodium Hydroxide | No   | No       | Yes        | Yes   | No      |
| Sodium Chloride  | No   | No       | No         | Yes   | No      |

#### **Europe**

**EEC Regulatory** 

All intentional ingredients are listed on the European EINECS Inventory.

# **SECTION 16 - OTHER INFORMATION**

#### **Revisional Updates**

5/29/2015 - Updated Sections 2.1 and 3.2 10/9/2013 - Released Version 1.0

#### **NFPA Codes**

Health 3 Flammability 0 Reactivity 2

#### **Dangers**

#### Sodium Hydroxide

H290 - May be corrosive to metals.

H314 - Causes severe skin burns and eye damage.

## **Sodium Chloride**

None

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